Preparation and Thermoelectric Properties of p and n-type CoSb₃

T. Caillat*, A. Borshchevsky and J. -P. Fleurial

Jet Propulsion Laboratory/California Institute of Technology, 4800 Oak Grove Drive, MS 277/212,1'asaclena,CA911 09
**\alpha: (81 8) 354-4036 Fax: (818) 393-6951

ABSTRACT

For the first time, large single crystals of the compound CoSb₃ were grown from antimony rich melts by the Bridgman gradient freeze technique. CoSb₃ has the skutterudite crystal structure and a peritectic decomposition temperature of 873°C. Samples were characterized by X-ray diffractometry, microprobe analysis and density measurements. A Debye temperature of 306 K was calculated from speed of sound measurements, As-grown · CoSb₃ crystals show p-type conductivity. N-type samples were obtained over a wide range of doping level by addition of Tc. Seebeck coefficient, electrical resistivity, thermal conductivity and Hall effect measurements were performed between room temperature and ,500 °C. The results of these measurements are presented and discussed. Exceptional y high Hall mobilities were obtained on p-type samples which is consistent with the covalency of the skutterudite crystal structure. A maximum room temperature Hallmobility of about 3445 cm².V⁻¹.s⁻¹ was measured on a sample with a carrier concentration of 4 x 10¹⁷ cm⁻³. N-type samples were found to have substantially lower Hall mobilities in the range of temperature investigated, A bandgap of 0.55 eV was estimated from the high temperature electrical resistivity and Hall effect measurements. The hole and electron effective masses were also estimated and are presented. Based on the experimental data obtained, the potential of CoSb₃ as a thermoelectric material is discussed.

INTI<ODUCTION

CoSb₃ belongs to a large family of materials having the skutterudite structure. This structure is composed of a cubic lattice, space group *Im*3 and the unit cell contains 8 AB₃ groups. The lattice constant of CoSb₃ is 9.0385 Å and its peritectic decomposition temperature is 873 "C [1]. Some properties of this compound were previously investigated and it was found that CoSb₃ is a semiconductor and its bandgap was estimated to 0.5 CV from high temperature electrical resistivity measurements on n-type polycrystalline samples [2]. However, results obtained on small single crystals prepared by chemical vapor transport techniques showed that CoSb₃ had a metallic behavior. To clarify these contracditory results, we attempted to grow, for the first time, large crystals of CoSb₃ from

non-stoichiometric melts using the gradient freeze technique. The samples were characterized by microprobe analyses, density and X-ray measurements. A variety of electrical and thermal measurements were also performed on these samples over a wide range of temperature. The experimental data obtained show that CoSb₃ is indeed a semiconductor, The results of the measurements are presented and the potential of this compound as a thermoelectric material is discussed.

EXPERIMENTAL DETAILS

The Co-Sb phase diagram show that the growth of the compound can be initiated from Sb rich melts between 91 and -97 at. % Sb [1]. Samples were grown using the gradient freeze technique from melts with 93 at.% Sb in scaled quartz ampoules coated with graphite and with a pointed bottom. A temperature gradient of about 50 °C/cm was maintained at the growth interface and the growth rate was 0.7 °C/h [3]. Doping studies were also conducted by adding different amounts of Te in the melt. Typical ingots obtained after the growth were composed of two parts: the bottom part corresponding to the compound CoSb₃ and , the upper part corresponding to the Sb rich cutcctic. An overall number of 20 samples in the form of disks about 2 mm thick and 10 mm in diameter were cut from the ingots. Typical samples were composed of a few very large grains but single crystals (-10 mm in diameter, 2 mm thick) were obtained as indicated by Lauc patterns. Selected samples were polished and their microstructure was investigated under an optical microscope. Microprobe analyses were also performed on the same samples to check their composition. Finally, some samples were ground for X-ray diffractometry (XRD) analyses. XRD experiments showed that the samples were single phase and the lattice constant calculated from the pattern showed that the stoichiometry of the samples was 1:3. The density of all samples was measured by the immersion technique using toluene as liquid, The measured densities were found to be about 99.5% of the theoretical density (7.69 g.cm⁻³) which also indicates that the stoichiometry of the samples was 1:3.

Samples were characterized at room temperature by speed of sound, Hall effect and Seebeck coefficient measurements. High temperature Seebeck coefficient, electrical resistivity, thermal conductivity and Hall effect measurements were also performed on selected samples,

RESULTS AND DISCUSSION

The velocity of sound was measured at room temperature in a single crystal of $CoSb_3$ about 8 mJn long and using a frequency of 5 MHz. The longitudinal sound velocity was found to be $4.62 \times 10^3 \, \text{m.s}^{-1}$ and the transversal $2.64 \times 10^3 \, \text{m.s}^{-1}$. The calculated average sound

velocity is 2,93 x 10³ m.s⁻¹ and the calculated Debye temperature using this average sound velocity is 306 K.

Room temperature van dcr Paw, Hall effect and Seebeck coefficient measurements arc summarized in Table 1 and the results clearly show that CoSb₃ is a semiconductor. As grown samples show p-type conductivity while doping with Tc concentration between 0.08 and 0.15 at. % changed the conductivity to n-type. All p-type samples were cut from the same ingot and are single crystals. The change in properties from one end of the ingot to the other might be attributed to changes in the stoichiometry of the samples inherent of the growth process or segregation of residual impurities towards the end of the ingot. N-type samples cut from the same ingot have different carrier concentration because of the segregation of Tc during the directional crystallization.

Table I. Room temperature properties of n and p-type CoSb₃ samples.

Table 1. Room temperature properties of it and p-type Coso3 samples.										
Sample	Nominal	Conductivity	ρ	R_{H}	n/p	$\mu_{\rm H}$	α			
#	dopant	type								
	(at.%)									
1NB13	-	p	0.8412	1.735	0.360	2062.5	141			
2NB13	-	р	0.8117	1.605	0.389	1981.0	139			
3NB13	-	p	0.7367	1.429	0.437	1944.0	138			
4NB13	-	р	0.6997	1.299	0.481	1856.0	135			
5NB13	-	р	0.7353	1.379	0.453	1877.0	139			
6NB13	-	p	1.076	2.316	0.270	2153.3	159			
71NB13	-	p	1.894	5.372	0.116	2835.7	189			
72NB13	-	р	4.523	15.580	0.040	3445.5	233			
1CS10	0.08 Te	n	14.2	-1.374	0.454	-101.0	-500			
3CS10	0.08 Te	n	7.627	-0.644	0.969	-84.4	-380			
ICS7	0.1 Te	n	11.64	-1.266	0.494	-108.8	-450			
2CS9	0.12 Te	n	5.99	-0.408	1.520	-68.0	-270			
2CS9A	0.12 Te	n	9.2	-0.620	0.920	-70.0	-410			
1CS11	0.15 Te	n	5.572	-0.485	1.287	-87.0	-373			

Electrical resistivity p (m Ω .cm), Hall coefficient R_H (C.cm⁻³), Hall carrier concentration n/p (x 10^{19} .cm⁻³), Hall mobility μ_H (cm².V⁻¹.s⁻¹), Seebeck coefficient a (pV/K).

P-type samples have exceptionally high carrier mobility and a maximum of 3445 cm².V⁻¹.s⁻¹ was measured on a sample with a Hall carrier concentration of about 4 x 10¹⁷ cm⁻³. Figure 1 shows the room temperature Hall mobility for p-type skutterudite compounds CoSb₃ and IrSb₃ and also for p-type Si and Gc.Skutterudite compounds have significantly higher mobility than state-of-the-art semiconductors such as Si, Gc and Ill-V's. These high mobilities are consistent with the covalent bonds in the skutterudite structure and also to the fact that the valence band is essentially derived from pnicogen-pnicogen bonds in these materials [4]. N-type mobilities are substantially lower resulting in higher electrical resistivity values. Large Seebeck coefficients were measured on n-type samples. An estimation of the effective masses can be made from the room temperature values of the

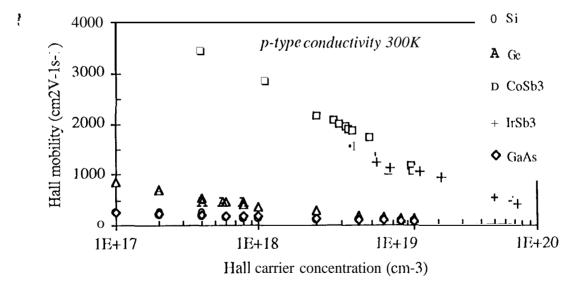
1 Iall coefficient and Scebeck coefficient as follows. The Scebeck coefficient (et) can be expressed using Fistul formalism as:

$$\alpha = \pm \frac{k}{c} \left(\frac{\Phi_4(\xi, b)}{\Phi_3(\xi, b)} - \xi \right)$$
 (1)

where k is the Boltzmann constant, c the electron charge, ξ the Fermi level, Φ_4 and Φ_3 are Fistul integrals and b is a factor which depends on the scattering mechanisms. Assuming that the dominant scattering mechanism is acoustic phonons, b=O. Using the same formalism, the Hall coefficient (R_H) can be expressed as:

$$R_{H_{\underline{}}} \pm \frac{3h3}{8\pi e(2m * kT)3 / 2} \frac{\Phi_{9,2}(\xi,b)}{[\Phi_{3}(\xi,b)]2}$$
 (2)

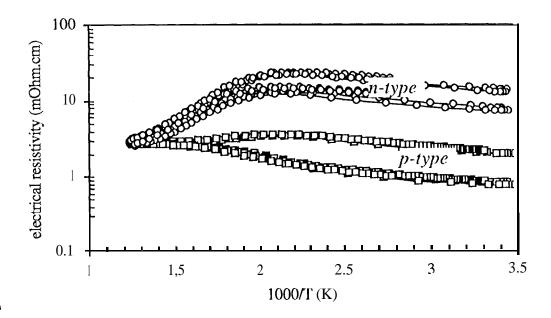
Using the room temperature values of the Seebeck coefficient and assuming acoustic phonon scattering, the Fermilevel ξ can be calculated from equation (1) and used in equation (2) to calculate the effective mass m^* . Using the data tabulated in table 1, an average mass of 0.16 and 1.65 m_o was calculated for the holes and the electrons, respectively.



<u>Figure 1:</u> Room temperature I Ian mobilit y for p-t ypc CoSb₃ and IrSb₃ as a function of Hall carrier concentration. Values for p-type Si and Ge arc also shown for comparison.

Figure 2 shows the electrical resistivity values as a function of the temperature for several n and p-type CoSb₃ samples with different doping levels. N-type samples have higher resistivity values than p-type which do not show any intrinsic behavior in the investigated range of temperature. Above 260°C, the variations of (he electrical resistivity of n-type

samples are practically linear with temperature and a bandgap value of 0.55 eV was estimated from the slopes of the curves.



<u>Figure 2:</u> Electrical resistivity as a function of temperature for p and n-type CoSb₃.

Figures 3 and 4 show the variations of the Hallmobility and Seebeck coefficient as a function of temperature for n and p-type CoSb₃ samples. The Hall mobility of p-type samples decreases with temperature as T-1.9 indicating that acoustic phonons is the dominant scattering mechanism. The Hall mobility of n-type samples is negative up to about 200°C and becomes positive for higher temperatures, increasing up to about 400°C and decreasing for higher temperatures. The Seebeck coefficient variations of n and p-type CoSb₃ samples with temperature are quantitatively similar to the variations of the Hall mobility. However, the Seebeck coefficient of n-type samples changes sign at a temperature of about 300°C, higher than for the Hall mobility, This unusual feature needs to be understood and will be adressed in the future.

The thermal conductivity of two p-type and one n-type CoSb₃ samples was calculated from thermal diffusivity and heat capacity measurements up to 700°C and the results arc shown in Figure 5. The thermal conductivity decreases from room temperature up to about 400°C, reaches a minimum value of about 4 W.m⁻¹.K⁻¹ and then increases due to increasing minority carrier conduction (bipolar contribution). The average value of 'the thermal conductivity at room temperature is 11.8 W.m⁻¹.K⁻¹. An estimation of the lattice thermal conductivity of CoSb₃ can be made by calculating the electronic contribution to the total thermal conductivity using the Wiedemann-Franz law, the value of the Lorenz number

being calculated using the Fermi level value obtained from equation (1). The lattice thermal conductivity (λ_{ph}) was estimated at 10.3 W.m⁻¹.K⁻¹.

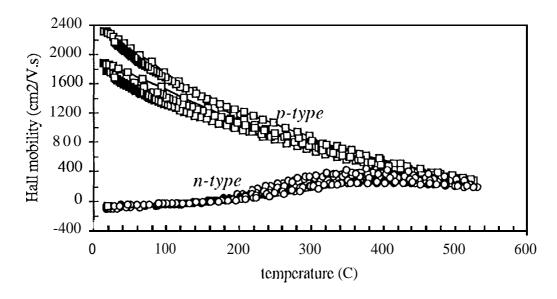


Figure 3: Hall mobility as a function of temperat urc for p and n-t ypc CoSb₃.

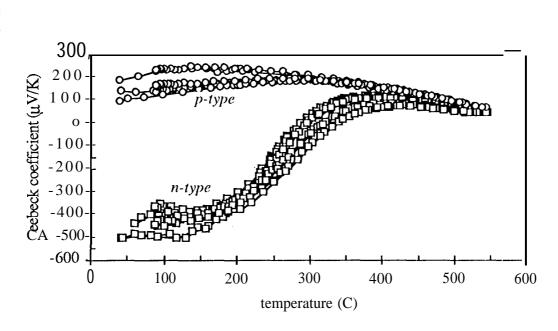
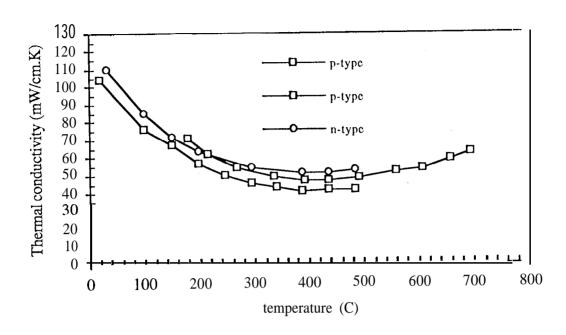


Figure 4: Seebeck coefficient as a function of temperature for p and n-type CoSb₃.



, Figure 5: Thermal conductivity as a function of temperature for p and n-type CoSb₃.

Figure 6 shows the calculated ZT values for two p-type samples and onc n-type samples as a function of temperature. A maximum ZT value of 0.15 was achieved at about 300°C. This value is relatively low but no efforts at this time were made to optimize the doping level of the samples.

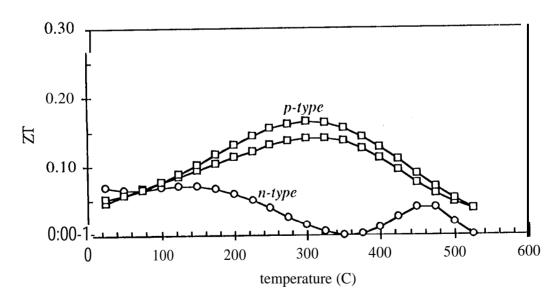


Figure 6: ZT values as a function of temperature for some p and n-type CoSb₃ samples.

Conventional theory for a single band semiconductor indicates the optimum thermoelectric figure of merit depends on the scattering mechanism and a material parameter $\beta = m^* 3/2.\mu/\lambda_{ph}$ where m^* is the carrier effective mass, μ is the carrier mobility and λ_{ph} is the lattice thermal conductivity. β can be calculated from the estimated electron and hole masses, the estimated lattice thermal conductivity of $10.3\,W.m^{-1}.K^{-1}$ and the measured values of the Hall nobilities for the holes and electrons. The Hall mobilities were extrapolated at a carrier concentration of $10^{17}\,cm^{-3}$ using the variations of the Hall mobility as a function of carrier concentration determined in this study. The results of the calculations are reported in Table 2.

Table 2. Some properties of n and p-type CoSb₃.

Tuble 2. Some properties of it and p type costs;								
	m*/m _o	$\mu ({\rm at} 10^{17} {\rm cm}^{-3})$	$\lambda_{ m ph}$	β				
		$(cm^2.V^{-1}.s^{-1})$	$(W.m^{-1}.K^{-1})$					
J)-type	0.16	5000	10.3	31.06				
n-t ypc	1.65	250	10.3	51.44				

• The results in Table 2 show that n-type samples might have higher ZT values than p-type despite the exceptionally Hall nobilities of p-type samples. CoSb₃ is definitively an interesting material with a good thermoelectric potential, However, relatively low ZT values, were obtained mainly because the lattice thermal conductivity of CoSb₃ is too large. High 'ZT values might be possible for material with the skutterudite structure if one can find a skutterudite material with a substantially lower thermal conductivity, retaining the good basic properties of CoSb₃. Considering the large number of materials having the skutterudite structure [5] and the possibilities of tailoring composition and band gap in these materials, one should be able to find a material with a good potential for high ZT values. Reduction of the thermal conductivity can be achieved, for example, by increasing phonon scattering by alloying isostructural compounds [6] or developing ternary or even quaternary skutterudite related phases [5],

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REFERENCES

- 1. **P.** Feschotte, D. Lorin, "Les Systèmes Binaires Fe-Sb, Co-Sb et Ni-Sb", Journal of he Less-Common Metals, 155, 25S (1989).
- 2. L. D. Dudkin, N. Kh. Abrikosov, "A Physicochemical Investigation of Cobalt Antimonides", Soviet Phys. Solid State, 3, 126(1959).
- 3. A. Borshchevsky, T. Caillat, J. -P. Fleurial, "*Two-Zone Bridgman Furnace With Sharp Thermal Gradient*", NASA Tech Briefs (Official Publication of the National Aeronautics and Space Administration), vol 18,3, pp 68-70, March 1994.
- 4. D. Jung, M. H. Whangbo, S. Alvarez, "Importance of the X4 Ring Orbitals for the Semicoducting, Metallic, or Superconducting Properties of Skutterudites MX₃ and RM₄X₁₂", Inorg. Chem., 29, 2252 (1 990).
- 5. T. Caillat, A. Borshchevsky, J. -P. Fleurial, "Existence and Some Properties of New Ternary Skutterudite Phases", this conference.
- 6. A. Borshchevsky, T. Caillat, J. -P. Fleurial, "CoSb₃-IrSb₃ Solid Solutions: Preparation and Characterization", this conference.